H3PO4 Acid is a handy tool for preparing recipes to mix, dilute, neutralize H3PO4 acid as well as get a quick safety and first aid directions

H3PO4 Home



The Home navigation screen provides access to all modules included in the app.

- 1. The "H3PO4 Dilution" screen allows the preparation of desired diluted acid volume from concentrated acid.
- 2. The "H3PO4 Neutralization" screen allows the calculation of the required quantity of various alkalis to neuttralize the given volume of H3PO4 acid.
- 3. The "Solution Preparation" screen allows the calculation of the required acid volume from a given concentration to mix a solution with a given "Molarity/Normality"
- 4. The "H3PO4 Properties" page gives a general information related to the product phisycal and chemical properties.

- 5. The "First Aid" page provide guidance to basic first aid steps to be taken when exposed to uncontrolled H3PO4 environment.
- 6. The "Settings" screen allows user to set his preferred units

H3PO4 Dilution



- 1. Return to Home screen
- 2. Go to Settings screen
- 3. Enter raw H3PO4 acid concentration
- 4. Enter desired H3PO4 acid concentration
- 5. Enter water density (water used for dilution)
- 6. Enter desired diluted acid volume
- 7. Output screen showing:
 - Initial acid density at 20 °C
 - Desired acid density at 20 °C
 - · Raw acid volume
 - Required water volume for diluting the acid

- 8. Name of saved calculation or type in a name to save calculation under (\approx)
- 9. Pickerview to revert back to a saved calculation
- 10. Save your current calculation under recommended name or user typed name
- 11. Swipe left to delete you selected record on the pickerview
- 12. Share the calculation

(\approx) each record has a case record name and a ID name. Typically, both are set by the app to be identical. If user wants to identify the record with a different record name that appears on the pickerview, please follow the below steps:

- 1. Complete all data entries (from step 1 to 6 above)
- 2. Tap anywhere on the screen to process calculation
- 3. Type the record name in textbox 8
- 4. Press save button (10)

H3PO4 Neutralization



- 1. Return to Home screen
- 2. Go to Settings screen
- 3. Enter H3PO4 acid concentration
- 4. Enter H3PO4 acid volume to be neutralized
- 5. Select the neutralization agent to be used (云)
- 6. The calculation output screen showing:
 - · Mass of the selected neutralizer used to neutralize a unit volume of acid
 - Total Mass of the selected neutralizer used to neutralize a entire volume of acid
- 7. The chemical reaction equation and complete name of neutralizer will appear here once the neutralizer is selceted in step (5)
- 8. Name of saved calculation or type in a name to save calculation under (**)
- 9. Pickerview to revert back to a saved calculation
- 10. Save your current calculation under recommended name or user typed name
- 11. Swipe left to delete you selected record on the pickerview
- 12. Share the calculation

 (\approx) some neutralizers have an extremely fast and exothermal reaction with a CO2 gas release when added to the acid. The commun neutralizers used on commercial basis are sodium bicarbonate, soda ash or lime. Safety considerations shall be well taken as per each product directives.

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- 3. Type the record name in textbox 8
- 4. Press save button (10)

H3PO4 Solution Preparation



- 1. Return to Home screen
- 2. Go to Settings screen
- 3. Enter H3PO4 acid concentration
- 4. Enter H3PO4 acid solution volume required
- 5. Select the solution unit measurment "Molarity or Normality"(1/2)
- 6. Enter the desired Molarity/Normality value
- 7. The calculation output screen showing:
 - Volume of raw acid needed
 - Volume of water needed
 - The equivalent acid concentration
 - The equivalent acid pH value
- 8. Name of saved calculation or type in a name to save calculation under (☆☆)
- 9. Pickerview to revert back to a saved calculation
- 10. Save your current calculation under recommended name or user typed name
- 11. Swipe left to delete you selected record on the pickerview
- 12. Share the calculation

(\approx) Normality is similar in concept to Molarity. Where molarity (M) represents the concentration of an ion or compound in solution, normality (N) represents the molar concentration of the acid component (usually the H+ ion in an acid solution). Typically N = (M)(number of hydrogen or hydroxide ions).

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- 2. Tap anywhere on the screen to process calculation
- 3. Type the record name in textbox 8
- 4. Press save button (10)

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🧠 📋 Solutio	n Preparati + Units
HSPO4 Acid Core.	70 5
HEPO4 Aski Valuese	1000 gif
Molarity	Normality
Desired Motority	0.1 u
Rev Acid Volume	9.18 gel
Water Volume	990.82 🕬
Final Concentration	1 %
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Bharing More	
See Image D	Crey Divis
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Share Screen

Share screen allows sharing the claulations screen via sharing apps available to each user

Settings



- 1. Go to Home screen
- 2. Select "Volume" unit
- 3. Select "Weight/mass" unit
- 4. Select 'Density" unit
- 5. Cick on "Save" to seve your units selection
- 6. the saved units are displayed in the "Saved Units" textbox to confirm proper selection